CITY OF REDLANDS MUNICIPAL UTILITIES & ENGINEERING DEPARTMENT TRIENNIAL PUBLIC HEALTH GOALS (PHGs) REPORT JULY 2020



1. INTRODUCTION

Provisions of the California Health and Safety Code (Attachment "A") specify that larger (> 10,000 service connections) water utilities prepare a special report if their water quality measurements have exceeded any Public Health Goals (PHGs). PHGs are non-enforceable goals established by the Cal-EPA's Office of Environmental Health Hazard Assessment (OEHHA). The law also requires that where OEHHA has not adopted a PHG for a constituent, the water suppliers are to use the MCLGs adopted by USEPA. Only constituents which have a California primary drinking water standard and for which either a PHG or MCLG has been set are to be addressed. Attachment "C" lists of all regulated constituents with the MCLs and PHGs or MCLGs.

There are a few constituents that are routinely detected in water systems at levels usually well below the drinking water standards for which no PHG nor MCLG has yet been adopted by OEHHA or USEPA including Total Trihalomethanes. These will be addressed in a future required report after a PHG has been adopted.

This law specifies what information is to be provided in the report (see Attachment "A").

If a constituent was detected in the City of Redlands water supply between 2017 and 2019 at a level exceeding an applicable PHG or MCLG, this report provides the information required by the law. Included is the numerical public health risk associated with the MCL and the PHG or MCLG, the category or type of risk to health that could be associated with each constituent, the best treatment technology available that could be used to reduce the constituent level, and an estimate of the cost to install that treatment if it is appropriate and feasible.

The purpose of the legislation requiring this report is to provide consumers with information on levels of contaminants even below the enforceable mandatory Maximum Contaminant Levels (MCLs) so consumers are aware of whatever risks might be posed by the presence of these contaminants at levels below the MCLs.

The Association of California Water Agencies (ACWA) formed a workgroup which prepared guidelines for water utilities to use in preparing these newly required reports. ACWA guidelines were used in the preparation of this public health goals report.

2. BACKGROUND

Public Health Goal (PHG)

PHGs are set by the California Office of Environmental Health Hazard Assessment (OEHHA) which is part of Cal-EPA and are based solely on public health risk considerations. None of the practical risk-management factors that are considered by the USEPA or the California Division of Drinking Water in setting drinking water standards (MCLs) are considered in setting the PHGs. These factors include analytical detection capability, treatment technology available, benefits and costs. The PHGs are not enforceable and are not required to be met by any public water system. MCLGs are the federal equivalent to PHGs.

Maximum Contaminant Level Goal (MCLG)

The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are non-enforceable public-health goals.

Maximum Contaminant Level (MCL)

The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to MCLGs as feasible using the best available treatment technology and taking cost into consideration. MCLs are enforceable standards for public water-supply systems.

The U.S. EPA and the California State Water Resources Control Board (SWRCB) Division of Drinking Water (DDW) establish MCLs at very conservative levels to provide protection to consumers against all but very low to negligible risk. In other words, MCLs are the regulatory definition of what is "safe." Adopted MCLs are still the criteria for being in compliance, not those proposed or possible in the future, and certainly not MCLGs or PHGs.

MCLGs and PHGs are often set at very low levels depending on the established health risk, and in the case of U.S. EPA, MCLGs are also set at zero for some contaminants. Determination of health risk at these low levels is theoretical based on risk assessments with multiple assumptions and mathematical extrapolations (see Attachment "D"). Many contaminants are considered to be carcinogenic and U.S. EPA's policy is to set the applicable MCLGs at zero because they consider no amount of these contaminants to be without risk. It is understood by all that zero is an unattainable goal and cannot be measured by the practically available analytical methods. Note that by regulation, OEHHA cannot set a PHG at zero and must calculate a numerical level to address risk, even though it may be unattainable or impossible to measure.

PHGs and MCLGs are not enforceable. The Best Available Technology (BAT) to reach such low levels has not been defined and may not realistically be available. Accurate cost estimates are difficult, if not impossible, and are highly speculative and theoretical. Therefore, they have limited value and may not warrant significant investment of agency time and money.

3. BEST AVAILABLE TREATMENT TECHNOLOGY AND COST ESTIMATES

Both the USEPA and DDW adopt what are known as BATs or Best Available Technologies which are the best known methods of reducing contaminant levels to the MCL. However, since many PHGs and all MCLGs are set much lower than the MCL, it is not always possible nor feasible to determine what treatment is needed to further reduce a constituent downward to or near the PHG or MCLG, many of which are set at zero. Estimating the costs to reduce a constituent to zero is difficult, if not impossible because it is not possible to verify by analytical means that the level has been lowered to zero. In some cases, installing treatment to try and further reduce very low levels of one constituent may have adverse effects on other aspects of water quality.

Costs estimates listed in the tables below for such technologies are based on the ACWA PHG survey of 2012 and internal historical cost models. The estimates are based on \$/1000 gallons of treated water which includes annualized capital and O&M costs.

Cost estimate formula =

ACWA estimate (\$/1000 gal) x annual historical usage per year / # of service connections

4. WATER QUALITY DATA

All of the water quality data collected by our water system between 2017 and 2019 for purposes of determining compliance with drinking water standards was considered. This data was all summarized in our 2017, 2018, and 2019 Consumer Confidence Reports which were mailed to all of our customers in June of each year. For additional information on the City's pervious Consumer Confidence Reports go to https://www.cityofredlands.org/post/water-quality.

5. CONSTITUENTS DETECTED THAT EXCEED A PHG OR MCLG

Table 1 identifies the constituents that were detected by the City of Redlands above the PHG or MCLG during the reporting period 2017 – 2019. Over the last three years the City of Redlands drinking water has met all MCLs and safe drinking water standards adopted by USEPA and DDW; however, three contaminates (Perchlorate, Strontium 90 and Uranium) exceeded PHGs, and three (Gross Alpha Particle Activity, Gross Beta Particle Activity and Total Coliform Bacteria) exceeded MCLGs. The following is a discussion of these constituents that were detected in one or more of our drinking water sources. For information related to the numerical health risks associated with each constituents listed see Attachment "D".

Table 1					
	Constituents detected above the PHG or MCLG (2017 - 2019)				
Constituent	PHG (MCLG)	MCL			
Total Coliform Bacteria	(0%)	5%			
Perchlorate	1 ug/L	6 ug/L			
Gross Alpha Particle Activity (pCi/L)	(0)	15 (pCi/L)			
Gross Beta Particle Activity (pCi/L)	(0)	50 (pCi/L)			
Strontium-90 (pCi/L)	0.35 (pCi/L)	8 (pCi/L)			
Uranium (pCi/L)	0.43 (pCi/L)	20 (pCi/L)			

Total Coliform Bacteria

The City of Redlands water system is required to sample for a minimum of 80 total coliform samples per month. Each month 80 to 130 samples are collected to monitor for total coliform bacteria. Over the reporting period each year results exceeded the MCLG of 0% however, no MCL violations occurred as a result.

The MCL for coliform is 5% positive samples of all samples per month and the MCLG is zero. The reason for the coliform drinking water standard is to minimize the possibility of the water containing pathogens which are organisms that cause waterborne disease. Because coliform is only a surrogate indicator of the potential presence of pathogens, it is not possible to state a specific numerical health risk. While USEPA normally sets MCLGs "at a level where no known or anticipated adverse effects on persons would occur", they indicate that they cannot do so with coliform bacteria.

Coliform bacteria is an indicator organism that are ubiquitous in nature and are not generally considered harmful. They are used because of the ease in monitoring and analysis. If a positive sample is found, it indicates a potential problem that needs to be investigated and follow up sampling performed. It is not at all unusual for a system to have an occasional positive sample. It is difficult, if not impossible, to assure that a system will never get a positive sample.

The City of Redlands adds chlorine at our sources to assure that the water served is microbiologically safe. The chlorine residual levels are carefully controlled to provide the best health protection without causing the water to have undesirable taste and odor or increasing the disinfection byproduct levels. This careful balance of treatment processes is essential to continue supplying our customers with safe drinking water.

Other equally important measures that we have implemented include: an effective cross-connection control program, maintenance of a disinfectant residual throughout our system, an effective monitoring and surveillance program and maintaining positive pressures in our distribution system. Our system has already taken all of the steps described by DDW as "best available technology" for coliform bacteria in Section 64447, Title 22, CCR (see Attachment "B" Article 12. Best available technologies (BAT)).

		7	Table 2			
	Months with one or more Total Coliform positive sample(s) (2017 - 2019)					
	2017 2018 2019					
Month	% Positive	Month	% Positive	Month	% Positive	
May	2.5%	October	2%	January	1%	
				March	1%	
				April	1%	

Perchlorate

Perchlorate is an oxidizing chemical used in a variety of industrial processes. Perchlorate can occur in the environment either through industrial contamination or from natural sources. Perchlorate is also an inorganic chemical used in solid rocket propellant, fireworks, explosives, flares, matches, and a variety of industries. It usually gets into drinking water as a result of environmental contamination from historic aerospace or other industrial operations that used or use, store, or dispose of perchlorate and its salts. Perchlorate exposure in the U.S. is ubiquitous, mostly from ingestion of perchlorate in contaminated food or water.

The numerical public health risk associated with Perchlorate is listed in Attachment "D" of this report. Note, the numerical cancer risk at the PHG and MCL are listed N/A = not applicable, cancer risk cannot be calculated.

The City of Redlands currently uses a DDW approved and permitted blending strategy to comply with the regulatory MCL standard. Table 3 lists the source well affected by the perchlorate PHG exceedance. The table lists the treatment strategy used for regulatory compliance and the proposed BAT for additional Perchlorate reductions to meet the PHG target.

The preferred BAT considered for well 38 is Ion Exchange (IX). Staff has evaluated the initial capital and O&M costs based off an existing IX treatment system currently in operation within the City's water system and the ACWA cost references. IX has proven to treat Perchlorate to levels at and below the PHG of 1 ug/L.

		Table	3			
	Perchlorate (2017 - 2019)					
Source	Perchlorate Concentration (ug/L)	Best Available Technology	* Current Treatment Strategy / (Proposed BAT Treatment)	Estimate Aggregate Annual Cost (\$/Year)	Estimate Annual Customer Cost (\$/Year)	
Well 38	4	5, 12	*Blend / (5)	\$ 574,000	\$ 25	

5 = Ion Exchange12 = Biological Fluidized Bed Reactor

Perchlorate MCL = 6 ug/L Perchlorate PHG = 1 ug/L

Detectable Level for the purpose of Reporting (DLR) = 4 ug/L

Gross Alpha Particles Activity

Gross Alpha Particle Activity is a screening test that is performed to measure the overall radioactivity in drinking water. Gross alpha measurements can indicate the presence of a number of alpha emitting radionuclides, such as uranium and radium. Gross alpha particles in water supplies are predominantly from erosion of natural deposits.

The Office of Environmental Health Hazard Assessment (OEHHA) has not established a PHG for gross alpha and the USEPA has set a MCLG of zero pCi/L due to the classification of gross alpha radioactivity as a carcinogenic.

The gross alpha MCL is 15 pCi/Land MCLG is list at zero. The category of health risk for gross alpha particles is carcinogenicity. Carcinogenic risk means capable of producing cancer. The numerical health risk based on USEA's MCLG is zero therefore the cancer risk is zero at the MCLG. OEHHA indicates the numerical public health cancer risk at the MCL of 15 pCi/L to 1×10^{-3} for a lifetime of exposure. This means one excess case of cancer per 1,000 people exposed for a lifetime

The BAT for Gross Alpha is identified as Reverse Osmosis. Table 4 lists the sources that exceeded the MCLG of zero and costs estimates for reverse osmosis treatment.

Table 4						
Gross Alpha Particles (2017 - 2019)						
Source	Gross Alpha Concentration (pCi/L)	Best Available Technology	* Current Treatment Strategy / (Proposed BAT Treatment)	Estimate Aggregate Annual Cost (\$/Year)	Estimate Annual Customer Cost (\$/Year)	
Airport 1	4	7	(7)	\$ 1,451,250	\$ 63	
Airport 2	8	7	(7)	\$ 187,500	\$ 8	
East Lugonia 3	8	7	(7)	\$ 412,500	\$ 18	
East Lugonia 6	7	7	(7)	\$ 243,750	\$ 11	
Madeira	6	7	(7)	\$ 1,050,000	\$ 46	
Maguet 2	7	7	(7)	\$ 412,500	\$ 18	
Mentone 2	12	7	(7)	\$ 750,000	\$ 33	
N. Orange 2	5	7	(7)	\$ 1,875,000	\$ 82	
Orange Street	15	7	*Blend / (7)	\$ 1,125,000	\$ 49	
S.B. Ave Well	6	7	(7)	\$ 562,500	\$ 24	
Well 39	8	7	(7)	\$ 562,500	\$ 24	
HWTP	4	7	*2 / (7)	\$ 7,350,000	\$ 320	
TWTP	4	7	*2 / (7)	\$ 5,250,000	\$ 228	

2 = coagulation/filtration

7 = Reverse Osmosis

Gross Alpha MCL = 15 pCi/L

Gross Alpha MCLG = 0 pCi/L

Detectable Level for the purpose of Reporting (DLR) = 3 pCi/L

Gross Beta Particle Activity

Gross beta particle activity is a measure of the total amount of radioactivity in a water sample attributable to the radioactive decay of beta-emitting elements. Gross beta particles in water supplies are predominantly from erosion of natural deposits.

The Office of Environmental Health Hazard Assessment (OEHHA) has not established a PHG for gross beta and the USEPA has set a MCLG of zero pCi/L due to the classification of gross beta radioactivity as a carcinogenic.

The gross beta MCL is 50 pCi/L and MCLG is list at zero. The category of health risk for gross beta particles is carcinogenicity. Carcinogenic risk means capable of producing cancer. The numerical public health cancer risk based on USEA's MCLG is zero therefore the cancer risk is zero at the MCLG. At the MCL of 50 pCi/L the numerical public health cancer risk is at $2x10^{-3}$. This means two excess case of cancer per 1,000 people exposed for a lifetime.

The BAT for gross beta is identified as reverse osmosis (RO) and ion exchange (IX). Table 5 lists all sources which result in an exceedance of the MCLG of zero and cost estimates for IX treatment.

	Table 5						
	Gross Beta Particles (2017 - 2019)						
Source	Gross Beta Concentration (pCi/L)	Best Available Technology	* Current Treatment Strategy / (Proposed BAT Treatment)	Estimate Aggregate Annual Cost (\$/Year)	Estimate Annual Customer Cost (\$/Year)		
Airport 1	4	5, 7	(5)	\$ 464,400	\$ 20		
East Lugonia 6	5	5, 7	(5)	\$ 78,000	\$ 3		
Mentone 2	13	5, 7	(5)	\$ 240,000	\$ 10		
N. Orange 1	8	5, 7	(5)	\$ 858,000	\$ 37		
Rees	5	5, 7	*5 / (5)	\$ 360,000	\$ 16		
S.B. Ave Well	10	5, 7	(5)	\$ 180,000	\$ 8		
Well 38	6	5, 7	(5)	\$ 576,000	\$ 25		

7 = Reverse Osmosis

5 = Ion Exchange

Gross Beta MCL = 50 pCi/L Gross Beta MCLG = 0 pCi/L

Detectable Level for the purpose of Reporting (DLR) = 4 pCi/L

• Strontium-90

The radionuclide, strontium-90 (90Sr), is created during the process of nuclear fission, which occurs in a nuclear reactor or an atomic explosion. Although 90Sr does not occur naturally, it is ubiquitous because of the worldwide atmospheric testing of nuclear weapons. These tests, which occurred during the 1950s and 60s, resulted in the deposition of 90Sr onto all continents and water bodies. As such, there is a small amount of 90Sr in most environmental media, including drinking water. 90Sr decays by emitting a beta particle.

The PHG of strontium-90 is 0.35 pCi/ and the MCL is listed at 8 pCi/L. The category of health risk for strontium-90 particles is carcinogenicity. Carcinogenic risk means capable of producing cancer. The numerical public health cancer risk at the PHG is 1×10^{-6} which means one excess case of cancer per 1,000,000 people exposed for a lifetime. At the MCL of 8 pCi/L the numerical public health cancer risk is 2×10^{-5} which means two excess cases of cancer per 100,000 people exposed for a lifetime.

Table 6 lists the source which result in an exceedance of the PHG of .35 pCi/L and cost estimates for IX treatment.

Table 6 Strontium-90 (2017 - 2019)					
Source	Strontium-90 Concentration (pCi/L)	Best Available Technology	* Current Treatment Strategy / (Proposed BAT Treatment)	Estimate Aggregate Annual Cost (\$/Year)	Estimate Annual Customer Cost (\$/Year)
Airport 1	2	5, 7	(5)	\$ 464,400	\$ 20

5 = lon Exchange

Strontium-90 MCL = 8 pCi/L

7 = Reverse Osmosis

Strontium-90 PHG = 0.35 pCi/L

Detectable Level for the purpose of Reporting (DLR) = 2 pCi/L

Uranium

Uranium is a naturally occurring radioactive element that is ubiquitous in the earth's crust. Uranium is found in ground and surface waters due to its natural occurrence in geological formations. Uranium particles in water supplies are predominantly from erosion of natural deposits.

The MCL for uranium is 20 pCi/L and the PHG for uranium is 0.43 pCi/L. The category of health risk for uranium particles is carcinogenicity. Carcinogenic risk means capable of producing cancer. The numerical public health cancer risk at the PHG is 1×10^{-6} which means one excess case of cancer per 1,000,000 people exposed for a lifetime. At the MCL of 20 pCi/Lthe numerical public health cancer risk is 5×10^{-5} which means five excess cases of cancer per 500,000 people exposed for a lifetime.

Table 7 lists the source which result in an exceedance of the PHG of .43 pCi/L and cost estimates for IX and enhances coagulation/filtration treatment.

	Table 7						
	Uranium						
Source	Uranium Concentration (pCi/L)	Best Available Technology	* Current Treatment Strategy / (Proposed BAT Treatment)	Estimate Aggregate Annual Cost (\$/Year)	Estimate Annual Customer Cost (\$/Year)		
Airport 1	3	2, 5, 6, 7	(5)	\$ 464,400	\$ 20		
Airport 2	4	2, 5, 6, 7	(5)	\$ 60,000	\$ 3		
East Lugonia 3	2	2, 5, 6, 7	(5)	\$ 132,000	\$ 6		
East Lugonia 6	3	2, 5, 6, 7	(5)	\$ 78,000	\$ 3		
Madeira	5	2, 5, 6, 7	(5)	\$ 336,000	\$ 15		
Maguet 2	3	2, 5, 6, 7	(5)	\$ 132,000	\$ 6		
N. Orange 1	2	2, 5, 6, 7	(5)	\$ 858,000	\$ 37		
N. Orange 2	4	2, 5, 6, 7	(5)	\$ 600,000	\$ 26		
Orange Street	12	2, 5, 6, 7	* Blend / (5)	\$ 360,000	\$ 16		
Rees	2	2, 5, 6, 7	(5)	\$ 360,000	\$ 16		
S.B. Ave Well	5	2, 5, 6, 7	(5)	\$ 180,000	\$ 8		
HWTP	4	2, 5, 6, 7	(2)	\$ 1,470,000	\$ 64		

2 = coagulation/filtration

5 = Ion Exchange

6 = lime softening 7 = Reverse Osmosis

Uranium PHG = .43 pCi/L Detectable Level for the purpose of Reporting (DLR) = 1 pCi/L

Uranium MCL = 20 pCi/L

6. RECOMMENDATIONS FOR FURTHER ACTION

The drinking water quality of the City of Redlands meets all State of California, DDW and USEPA drinking water standards set to protect public health. To further reduce the levels of the constituents identified in this report that are already significantly below the health-based Maximum Contaminant Levels established to provide "safe drinking water", additional costly treatment processes would be required. The effectiveness of the treatment processes to provide any significant reductions in constituent levels at these already low values is uncertain. The health protection benefits of these further hypothetical reductions are not at all clear and may not be quantifiable. Therefore, no action is proposed.

7. REFERENCES

- Attachment "A" Excerpt from California Health & Safety Code: Section 116470 (b)
- Attachment "B" Excerpts from California Regulations Related to Drinking Water, Title 22 California Code of Regulations
- Attachment "C" Table of Regulated Constituents with MCLs, PHGs or MCLGs
- Attachment "D" Health Risk Information for Public Health Goal Exceedance Reports

ATTACHMENT "A"

Excerpt from California Health & Safety Code Section 116470 (b)

116470 (b) On or before July 1, 1998, and every three years thereafter, public water systems serving more than 10,000 service connections that detect one or more contaminants in drinking water that exceed the applicable public health goal, shall prepare a brief written report in plain language that does all of the following:

- (1) Identifies each contaminant detected in drinking water that exceeds the applicable public health goal.
- (2) Discloses the numerical public health risk, determined by the office, associated with the maximum contaminant level for each contaminant identified in paragraph (1) and the numerical public health risk determined by the office associated with the public health goal for that contaminant.
- (3) Identifies the category of risk to public health, including, but not limited to, carcinogenic, mutagenic, teratogenic, and acute toxicity, associated with exposure to the contaminant in drinking water, and includes a brief plainly worded description of these terms.
- (4) Describes the best available technology, if any is then available on a commercial basis, to remove the contaminant or reduce the concentration of the contaminant. The public water system may, solely at its own discretion, briefly describe actions that have been taken on its own, or by other entities, to prevent the introduction of the contaminant into drinking water supplies.
- (5) Estimates the aggregate cost and the cost per customer of utilizing the technology described in paragraph (4), if any, to reduce the concentration of that contaminant in drinking water to a level at or below the public health goal.
- (6) Briefly describes what action, if any, the local water purveyor intends to take to reduce the concentration of the contaminant in public drinking water supplies and the basis for that decision.
- (c) Public water systems required to prepare a report pursuant to subdivision (b) shall hold a public hearing for the purpose of accepting and responding to public comment on the report. Public water systems may hold the public hearing as part of any regularly scheduled meeting.
- (d) The department shall not require a public water system to take any action to reduce or eliminate any exceedance of a public health goal.
- (e) Enforcement of this section does not require the department to amend a public water system's operating permit.
- (f) Pending adoption of a public health goal by the Office of Environmental Health Hazard Assessment pursuant to subdivision (c) of Section 116365, and in lieu thereof, public water systems shall use the national maximum contaminant level goal adopted by the United States Environmental Protection Agency for the corresponding contaminant for purposes of complying with the notice and hearing requirements of this section.

ATTACHMENT "B"

Excerpt from California Regulations Related to Drinking Water

TITLE 22 CODE OF REGULATIONS

§64400.32. Detected.

"Detected" means at or above the detection limit for purposes of reporting (DLR).

§64400.34. Detection Limit for Purposes of Reporting (DLR).

"Detection limit for purposes of reporting (DLR)" means the designated minimum level at or above which any analytical finding of a contaminant in drinking water resulting from monitoring required under this chapter shall be reported to the State Board.

§64400.70. MCL.

"MCL" means maximum contaminant level.

§64401.80. Total Coliform-positive.

"Total coliform-positive" means a sample result in which the presence of total coliforms has been demonstrated.

Article 12. Best available technologies (BAT)

§64447. Best Available Technologies (BAT) – Microbiological Contaminants.

The technologies identified by the State Board as the best available technology, treatment techniques, or other means available for achieving compliance with the total coliform MCL are as follows:

- (a) Protection of wells from coliform contamination by appropriate placement and construction;
- (b) Maintenance of a disinfectant residual throughout the distribution system;
- (c) Proper maintenance of the distribution system; and
- (d) Filtration and/or disinfection of approved surface water, in compliance with Section 64650, or disinfection of groundwater

§64447.2. Best Available Technologies (BAT) - Inorganic chemicals.

The technologies listed in table 64447.2-A are the best available technology, treatment techniques, or other means available for achieving compliance with the MCLs in table 64431-A for inorganic chemicals.

Table 64447.2-A Best Available Technologies (BAT) Inorganic Chemicals

Chemical	Best Available Technologies (BATs)
Aluminum	10
Antimony	2, 7
Arsenic	1, 2, 5, 6, 7, 9, 13
Asbestos	2, 3, 8
Barium	5, 6, 7, 9
Beryllium	1, 2, 5, 6, 7
Cadmium	2, 5, 6, 7
Chromium	$2, 5, 6^{a}, 7$
Cyanide	5, 7, 11
Fluoride	1
Mercury	2^{b} , 4, 6^{b} , 7^{b}
Nickel	5, 6, 7
Nitrate	5, 7, 9
Nitrite	5, 7
Perchlorate	5,12
Selenium	$1, 2^{c}, 6, 7, 9$
Thallium	1, 5

^aBAT for chromium III (trivalent chromium) only.

Key to BATs in table 64447.2:

- 1 = Activated Alumina
- 2 = Coagulation/Filtration (not BAT for systems < 500 service connections)
- 3 = Direct and Diatomite Filtration
- 4 = Granular Activated Carbon
- 5 = Ion Exchange
- 6 = Lime Softening (not BAT for systems < 500 service connections)
- 7 = Reverse Osmosis
- 8 = Corrosion Control
- 9 = Electrodialysis
- 10 = Optimizing treatment and reducing aluminum added
- 11 = Chlorine oxidation
- 12 = Biological fluidized bed reactor
- 13 = Oxidation/Filtration

^bBAT only if influent mercury concentrations <10 ug/L.

^cBAT for selenium IV only.

§64447.3. Best Available Technologies (BAT) - Radionuclides.

The technologies listed in tables 64447.3-A, B and C are the best available technology, treatment technologies, or other means available for achieving compliance with the MCLs for radionuclides in tables 64442 and 64443.

Table 64447.3-A Best Available Technologies (BATs) Radionuclides

Radionuclide	Best Available Technology
Combined radium-226 and radium-228	Ion exchange, reverse osmosis, lime softening
Uranium	Ion exchange, reverse osmosis, lime softening, coagulation/filtration
Gross alpha particle activity	Reverse osmosis
Beta particle and photon radioactivity	Ion exchange, reverse osmosis

Table 64447.3-B
Best Available Technologies (BATs) and Limitations for Small Water Systems
Radionuclides

Unit Technologies	Limitations (see	Operator Skill Level	Raw Water Quality Range and Considerations
	footnotes)	Required	Considerations
1. Ion exchange	(a)	Intermediate	All ground waters; competing anion concentrations may affect regeneration frequency
2. Point of use, ion exchange	(b)	Basic	All ground waters; competing anion concentrations may affect regeneration frequency
3. Reverse osmosis	(c)	Advanced	Surface waters usually require pre- filtration
4. Point of use, reverse osmosis	(b)	Basic	Surface waters usually require pre- filtration
5. Lime softening	(d)	Advanced	All waters
6. Green sand filtration	(e)	Basic	All ground waters; competing anion concentrations may affect regeneration frequency
7. Co-precipitation with barium sulfate	(f)	Intermediate to advanced	Ground waters with suitable quality
8. Electrodialysis/electrodialysis reversal	(g)	Basic to intermediate	All ground waters

9. Pre-formed hydrous manganese oxide filtration	(h)	Intermediate	All ground waters
10. Activated alumina	(a), (i)	Advanced	All ground waters; competing anion concentrations may affect regeneration frequency
11. Enhanced coagulation/filtration	(j)	Advanced	Can treat a wide range of water qualities

Limitation Footnotes:

- ^a The regeneration solution contains high concentrations of the contaminant ions, which could result in disposal issues.
- ^b When point of use devices are used for compliance, programs for long-term operation, maintenance, and monitoring shall be provided by systems to ensure proper performance.
- ^c Reject water disposal may be an issue.
- ^d The combination of variable source water quality and the complexity of the water chemistry involved may make this technology too complex for small systems.
- ^e Removal efficiencies can vary depending on water quality.
- f Since the process requires static mixing, detention basins, and filtration, this technology is most applicable to systems with sufficiently high sulfate levels that already have a suitable filtration treatment train in place.
- g Applies to ionized radionuclides only.
- ^h This technology is most applicable to small systems with filtration already in place.
- ⁱ Chemical handling during regeneration and pH adjustment may be too difficult for small systems without an operator trained in these procedures.
- ^j This would involve modification to a coagulation/filtration process already in place.

Table 64447.3-C Best Available Technologies (BATs) for Small Water Systems by System Size Radionuclides

Compliance Technologies for System Size Categories Based On Population Served						
	25-500	501-3,300	3,301 - 10,000			
		Unit Technologies				
Contaminant	(Numbers Correspond to Table 64447.3-B)					
Combined radium-226 and radium-228	1, 2, 3, 4, 5, 6, 7, 8, 9 1, 2, 3, 4, 5, 6, 7, 8, 9 1, 2, 3, 4, 5, 6, 7, 8, 9					
Gross alpha particle activity	3, 4	3, 4	3, 4			
Beta particle activity and photon radioactivity	1, 2, 3, 4 1, 2, 3, 4 1, 2, 3, 4					
Uranium	1, 2, 4, 10, 11	1, 2, 3, 4, 5, 10, 11	1, 2, 3, 4, 5, 10, 11			

ATTACHMENT "C"

MCLs, DLRs, and PHGs for Regulated Drinking Water Contaminants (Units are in milligrams per liter (mg/L), unless otherwise noted.)

Last Update: March 13, 2019

This table includes:

For comparison:

California's maximum contaminant levels (MCLs)

Detection limits for purposes of reporting (DLRs)

<u>Public health goals (PHGs) from the Office of Environmental Health Hazard Assessment (OEHHA)</u>

Federal MCLs and
Maximum
Contaminant Level
Goals (MCLGs)
(US EPA)

Also, the PHG for NDMA (which is not yet regulated) is included at the bottom of this table.

Regulated Contaminant	MCL	DLR	PHG	Date of PHG	MCL	MCLG
Chemicals with MCLs in 22 Co						
Aluminum	1	0.05	0.6	2001		
Antimony	0.006	0.006	0.001	2016	0.006	0,006
Arsenic	0.010	0.002	0.000004	2004	0.010	zero
Asbestos (MFL = million fibers per liter; for fibers >10 microns long)	7 MFL	0.2 MFL	7 MFL	2003	7 MFL	7 MFL
Barium	1	0.1	2	2003	2	2
Beryllium	0.004	0.001	0.001	2003	0.004	0.004
Cadmium	0.005	0.001	0.00004	2006	0.005	0.005
Chromium, Total - OEHHA withdrew the 0.0025-mg/L PHG	0.05	0.01	withdrawn Nov. 2001	1999	0.1	0.1
Chromium, Hexavalent - 0.01-mg/L MCL & 0.001-mg/L DLR repealed September 2017			0.00002	2011		
Cyanide	0.15	0.1	0.15	1997	0.2	0.2
Fluoride	2	0.1	1	1997	4.0	4.0
Mercury (inorganic)	0.002	0.001	0.0012	1999 (rev2005)*	0.002	0.002
Nickel	0.1	0.01	0.012	2001		
Nitrate (as nitrogen, N)	10 as N	0.4	45 as NO3 (=10 as N)	2018	10	10
Nitrite (as N)	1 as N	0.4	1 as N	2018	1	1
Nitrate + Nitrite (as N)	10 as N		10 as N	2018		
Perchlorate	0.006	0.004	0.001	2015		
Selenium	0.05	0.005	0.03	2010	0.05	0.05
Thallium	0.002	0.001	0.0001	1999 (rev2004)	0.002	0.0005
Copper and Lead, 22 CCR §64672.3						
Values referred to as MCLs for lead and copper are not actually MCLs; instead, they are called "Action Levels" under the lead and copper rule						

Copper	1.3	0.05	0.3	2008	1.3	1.3
Lead	0.015	0.005	0.0002	2009	0.015	zero
Radionuclides with MCLs in 22 CCI	R §64441 an	nd §64443—	-Radioactivit	у		
[units are picocuries per liter (pCi/L), unle	ess otherwise	e stated; n/a	a = not applica	able]		
Gross alpha particle activity - OEHHA concluded in 2003 that a PHG was not practical	15	3	none	n/a	15	zero
Gross beta particle activity - OEHHA concluded in 2003 that a PHG was not practical	4 mrem/yr	4	none	n/a	4 mrem/yr	zero
Radium-226		1	0.05	2006		
Radium-228		1	0.019	2006		
Radium-226 + Radium-228	5				5	zero
Strontium-90	8	2	0.35	2006		
Tritium	20,000	1,000	400	2006		
Uranium	20	1	0.43	2001	30 ug/L	zero
Chemicals with MCLs in 22 CC	CR §64444—	Organic C	hemicals			
(a) Volatile Organic	: Chemicals	(VOCs)				
Benzene	0.001	0.0005	0.00015	2001	0.005	zero
Carbon tetrachloride	0.0005	0.0005	0.0001	2000	0.005	zero
1,2-Dichlorobenzene	0.6	0.0005	0.6	1997 (rev2009)	0.6	0.6
1,4-Dichlorobenzene (p-DCB)	0.005	0.0005	0.006	1997	0.075	0.075
1,1-Dichloroethane (1,1-DCA)	0.005	0.0005	0.003	2003		
1,2-Dichloroethane (1,2-DCA)	0.0005	0.0005	0.0004	1999 (rev2005)	0.005	zero
1,1-Dichloroethylene (1,1-DCE)	0.006	0.0005	0.01	1999	0.007	0.007
cis-1,2-Dichloroethylene	0.006	0.0005	0.013	2018	0.07	0.07
trans-1,2-Dichloroethylene	0.01	0.0005	0.05	2018	0.1	0.1
Dichloromethane (Methylene chloride)	0.005	0.0005	0.004	2000	0.005	zero
1,2-Dichloropropane	0.005	0.0005	0.0005	1999	0.005	zero
1,3-Dichloropropene	0.0005	0.0005	0.0002	1999 (rev2006)		
Ethylbenzene	0.3	0.0005	0.3	1997	0.7	0.7
Methyl tertiary butyl ether (MTBE)	0.013	0.003	0.013	1999		
Monochlorobenzene	0.07	0.0005	0.07	2014	0.1	0.1
Styrene	0.1	0.0005	0.0005	2010	0.1	0.1
1,1,2,2-Tetrachloroethane	0.001	0.0005	0.0001	2003	0.1	0.1
Tetrachloroethylene (PCE)	0.005	0.0005	0.00006	2001	0.005	zero
Toluene	0.15	0.0005	0.15	1999	1	1
1,2,4-Trichlorobenzene	0.005	0.0005	0.005	1999	0.07	0.07
1,1,1-Trichloroethane (1,1,1-TCA)	0.200	0.0005	1	2006	0.2	0.2
1,1,2-Trichloroethane (1,1,2-TCA)	0.005	0.0005	0.0003	2006	0.005	0.003
Trichloroethylene (TCE)	0.005	0.0005	0.0017	2009	0.005	zero
Trichlorofluoromethane (Freon 11)	0.15	0.005	1.3	2014		

1,1,2-Trichloro-1,2,2-Trifluoro ethane (Freon 113)	1.2	0.01	4	1997 (rev2011)		
Vinyl chloride	0.0005	0.0005	0.00005	2000	0.002	zero
Xylenes	1.750	0.0005	1.8	1997	10	10
(b) Non-Volatile Synthetic	Organic Ch	nemicals (S	OCs)			
Alachlor	0.002	0.001	0.004	1997	0.002	zero
Atrazine	0.001	0.0005	0.00015	1999	0.003	0.003
Bentazon	0.018	0.002	0.2	1999 (rev2009)		
Benzo(a)pyrene	0.0002	0.0001	0.000007	2010	0.0002	zero
Carbofuran	0.018	0.005	0.0007	2016	0.04	0.04
Chlordane	0.0001	0.0001	0.00003	1997 (rev2006)	0.002	zero
Dalapon	0.2	0.01	0.79	1997 (rev2009)	0.2	0.2
1,2-Dibromo-3-chloropropane (DBCP)	0.0002	0.00001	0.0000017	1999	0.0002	zero
2,4-Dichlorophenoxyacetic acid (2,4-D)	0.07	0.01	0.02	2009	0.07	0.07
Di(2-ethylhexyl)adipate	0.4	0.005	0.2	2003	0.4	0.4
Di(2-ethylhexyl)phthalate (DEHP)	0.004	0.003	0.012	1997	0.006	zero
Dinoseb	0.007	0.002	0.014	1997 (rev2010)	0.007	0.007
Diquat	0.02	0.004	0.006	2016	0.02	0.02
Endothal	0.1	0.045	0.094	2014	0.1	0.1
Endrin	0.002	0.0001	0.0003	2016	0.002	0.002
Ethylene dibromide (EDB)	0.00005	0.00002	0.00001	2003	0.00005	zero
Glyphosate	0.7	0.025	0.9	2007	0.7	0.7
Heptachlor	0.00001	0.00001	0.000008	1999	0.0004	zero
Heptachlor epoxide	0.00001	0.00001	0.000006	1999	0.0002	zero
Hexachlorobenzene	0.001	0.0005	0.00003	2003	0.001	zero
Hexachlorocyclopentadiene	0.05	0.001	0.002	2014	0.05	0.05
Lindane	0.0002	0.0002	0.000032	1999 (rev2005)	0.0002	0.0002
Methoxychlor	0.03	0.01	0.00009	2010	0.04	0.04
Molinate	0.02	0.002	0.001	2008		
Oxamyl	0.05	0.02	0.026	2009	0.2	0.2
Pentachlorophenol	0.001	0.0002	0.0003	2009	0.001	zero
Picloram	0.5	0.001	0.166	2016	0.5	0.5
Polychlorinated biphenyls (PCBs)	0.0005	0.0005	0.00009	2007	0.0005	zero
Simazine	0.004	0.001	0.004	2001	0.004	0.004
Thiobencarb	0.07	0.001	0.042	2016		
Toxaphene	0.003	0.001	0.00003	2003	0.003	zero
1,2,3-Trichloropropane	0.000005	0.000005	0.0000007	2009		
2,3,7,8-TCDD (dioxin)	3x10 ⁻⁸	5x10 ⁻⁹	5x10 ⁻¹¹	2010	3x10 ⁻⁸	zero
2,4,5-TP (Silvex)	0.05	0.001	0.003	2014	0.05	0.05
Chemicals with MCLs in 22 CCR		L	1			
Total Trihalomethanes	0.080				0.080	
Bromodichloromethane		0.0010	0.00006	2018 draft		zero

Bromoform		0.0010	0.0005	2018 draft		zero	
Chloroform		0.0010	0.0004	2018 draft		0.07	
Dibromochloromethane		0.0010	0.0001	2018 draft		0.06	
Haloacetic Acids (five) (HAA5)	0.060				0.060		
Monochloroacetic Acid		0.0020				0.07	
Dichloroacetic Acid		0.0010				zero	
Trichloroacetic Acid		0.0010			-	0.02	
Monobromoacetic Acid		0.0010					
Dibromoacetic Acid		0.0010					
Bromate	0.010	0.0050**	0.0001	2009	0.01	zero	
Chlorite	1.0	0.020	0.05	2009	1	0.8	
Chemicals with PHGs established in response			These are no	t currently			
I-Nitrosodimethylamine (NDMA)			0.000003	2006			
*OEHHA's review of this chemical during the year indicated (rev20XX) resulted in no change in the PHG.							

ATTACHMENT "D"

Public Health Goals

Health Risk Information for Public Health Goal Exceedance Reports

February 2019



Pesticide and Environmental Toxicology Branch Office of Environmental Health Hazard Assessment California Environmental Protection Agency

Health Risk Information for Public Health Goal Exceedance Reports

Prepared by

Office of Environmental Health Hazard Assessment California Environmental Protection Agency

February 2019

Under the Calderon-Sher Safe Drinking Water Act of 1996 (the Act), public water systems with more than 10,000 service connections are required to prepare a report every three years for contaminants that exceed their respective Public Health Goals (PHGs). This document contains health risk information on regulated drinking water contaminants to assist public water systems in preparing these reports. A PHG is the concentration of a contaminant in drinking water that poses no significant health risk if consumed for a lifetime. PHGs are developed and published by the Office of Environmental Health Hazard Assessment (OEHHA) using current risk assessment principles, practices and methods. ²

The water system's report is required to identify the health risk category (e.g., carcinogenicity or neurotoxicity) associated with exposure to each regulated contaminant in drinking water and to include a brief, plainly worded description of these risks. The report is also required to disclose the numerical public health risk, if available, associated with the California Maximum Contaminant Level (MCL) and with the PHG for each contaminant. This health risk information document is prepared by OEHHA every three years to assist the water systems in providing the required information in their reports.

Numerical health risks: Table 1 presents health risk categories and cancer risk values for chemical contaminants in drinking water that have PHGs.

The Act requires that OEHHA publish PHGs based on health risk assessments using the most current scientific methods. As defined in statute, PHGs for non-carcinogenic chemicals in drinking water are set at a concentration "at which no known or anticipated adverse health effects will occur, with an adequate margin of safety." For carcinogens, PHGs are set at a concentration that "does not pose any significant risk to health."

PHGs provide one basis for revising MCLs, along with cost and technological feasibility. OEHHA has been publishing PHGs since 1997 and the entire list published to date is shown in Table 1.

Table 2 presents health risk information for contaminants that do not have PHGs but have state or federal regulatory standards. The Act requires that, for chemical contaminants with California MCLs that do not yet have PHGs, water utilities use the federal Maximum Contaminant Level Goal (MCLG) for the purpose of complying with the requirement of public notification. MCLGs, like PHGs, are strictly health based and include a margin of safety. One difference, however, is that the MCLGs for carcinogens are set at zero because the US Environmental Protection Agency (US EPA) assumes there is no absolutely safe level of exposure to such chemicals. PHGs, on the other hand, are set at a level

¹ Health and Safety Code Section 116470(b)

² Health and Safety Code Section 116365

considered to pose no *significant* risk of cancer; this is usually no more than a one-in-one-million excess cancer risk (1×10^6) level for a lifetime of exposure. In Table 2, the cancer risks shown are based on the US EPA's evaluations.

For more information on health risks: The adverse health effects for each chemical with a PHG are summarized in a PHG technical support document. These documents are available on the OEHHA website (http://www.oehha.ca.gov). Also, technical fact sheets on most of the chemicals having federal MCLs can be found at http://www.epa.gov/your-drinking-water-contaminants.

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Chemical	Health Risk Category³	California PHG (mg/L) ⁴	Cancer Risk ⁵ at the PHG	California MCL ⁶ (mg/L)	Cancer Risk at the California MCL
Alachlor	carcinogenicity (causes cancer)	0.004	NA ⁷ ,6	0.002	NA
Aluminum	neurotoxicity and immunotoxicity (harms the nervous and immune systems)	0.6	NA	1	NA
Antimony	hepatotoxicity (harms the liver)	0.001	NA	0.006	NA
Arsenic	carcinogenicity (causes cancer)	0.000004 (4×10 ⁻⁸)	1×10 ⁻⁶ (one per million)	0.01	2.5×10 ⁻³ (2.5 per thousand)
<u>Asbestos</u>	carcinogenicity (causes cancer)	7 MFL ⁷ (fibers >10 microns in length)	1×10 ⁻⁶	7 MFL (fibers >10 microns in length)	1×10 ⁻⁶ (one per million)

³ Based on the OEHHA PHG technical support document unless otherwise specified. The categories are the hazard traits defined by OEHHA for California's Toxics Information Clearinghouse (online at: http://oehha.ca.gov/multimedia/green/pdf/GC Regtext011912.pdf).

⁴ mg/L = milligrams per liter of water or parts per million (ppm)

⁵ Cancer Risk = Upper bound estimate of excess cancer risk from lifetime exposure. Actual cancer risk may be lower or zero. 1×10⁻⁶ means one excess cancer case per million people exposed.

⁶ MCL = maximum contaminant level.

⁷ NA = not applicable. Cancer risk cannot be calculated.

⁸ The PHG for alachlor is based on a threshold model of carcinogenesis and is set at a level that is believed to be without any significant cancer risk to individuals exposed to the chemical over a lifetime. ⁷ MFL = million fibers per liter of water.

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

<u>Atrazine</u>	carcinogenicity (causes	0.00015	1×10 ⁻⁶	0.001	7×10-6
	cancer)				(seven per million)

Ol a start	Health Risk Category¹	California PHG	Cancer Risk ³ at the	California MCL ⁴	Cancer Risk at the
Chemical		(mg/L) ²	PHG	(mg/L)	California MCL
<u>Barium</u>	cardiovascular toxicity (causes high blood pressure)	2	NA	1	NA
<u>Bentazon</u>	hepatotoxicity and digestive system toxicity (harms the liver, intestine, and causes body weight effects ⁹)	0.2	NA	0.018	NA
<u>Benzene</u>	carcinogenicity (causes leukemia)	0.00015	1×10 ⁻⁶	0.001	7×10 ⁻⁶ (seven per million)
Benzo[a]pyrene	carcinogenicity (causes cancer)	0.000007 (7×10 ⁻⁶)	1×10-6	0.0002	3×10 ⁻⁵ (three per hundred thousand)
<u>Beryllium</u>	digestive system toxicity (harms the stomach or intestine)	0.001	NA	0.004	NA

⁹ Body weight effects are an indicator of general toxicity in animal studies.

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

<u>Bromate</u>	carcinogenicity (causes	0.0001	1×10 ⁻⁶	0.01	1×10-4
	cancer)				(one perten thousand)
Cadmium	nephrotoxicity	0.00004	NA	0.005	NA
	(harms the kidney)				
Carbofuran	reproductive toxicity (harms the testis)	0.0007	NA	0.018	NA

Chemical	Health Risk Category¹	California PHG (mg/L) ²	Cancer Risk³ at the PHG	California MCL⁴ (mg/L)	Cancer Risk at the California MCL
<u>Carbon</u> <u>tetrachloride</u>	carcinogenicity (causes cancer)	0.0001	1×10 ⁻⁶	0.0005	5×10 ⁻⁶ (five per million)
Chlordane	carcinogenicity (causes cancer)	0.00003	1×10 ⁻⁶	0.0001	3×10 ⁻⁶ (three per million)
<u>Chlorite</u>	hematotoxicity (causes anemia) neurotoxicity (causes neurobehavioral effects)	0.05	NA	1	NA
Chromium, hexavalent	carcinogenicity (causes cancer)	0.00002	1×10 ⁻⁶	none	NA

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Copper	digestive system toxicity (causes nausea, vomiting, diarrhea)	0.3	NA	1.3 (AL ¹⁰)	NA
<u>Cyanide</u>	neurotoxicity (damages nerves) endocrine toxicity (affects the thyroid)	0.15	NA	0.15	NA
<u>Dalapon</u>	nephrotoxicity (harms the kidney)	0.79	NA	0.2	NA
Di(2-ethylhexyl) adipate (DEHA)	developmental toxicity (disrupts development)	0.2	NA	0.4	NA
Diethylhexylphthalate (DEHP)	carcinogenicity (causes cancer)	0.012	1×10 ⁻⁶	0.004	3×10 ⁻⁷ (three per ten million)

Chemical	Health Risk Category¹	California PHG (mg/L) ²	Cancer Risk³ at the PHG	California MCL ⁴ (mg/L)	Cancer Risk at the California MCL
1,2-Dibromo- 3chloropropane (DBCP)	carcinogenicity (causes cancer)	0.000017 (1.7x10 ⁻⁶)	1×10 ⁻⁶	0.0002	1×10-4 (one perten thousand)

¹⁰ AL = action level. The action levels for copper and lead refer to a concentration measured at the tap. Much of the copper and lead in drinking water is derived from household plumbing (The Lead and Copper Rule, Title 22, California Code of Regulations [CCR] section 64672.3).

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

1,2-Dichlorobenzene (O-DCB)	hepatotoxicity (harms the liver)	0.6	NA	0.6	NA
1,4-Dichlorobenzene (p-DCB)	carcinogenicity (causes cancer)	0.006	1×10 ⁻⁶	0.005	8×10 ⁻⁷ (eight perten million)
1,1-Dichloroethane (1,1-DCA)	carcinogenicity (causes cancer)	0.003	1×10 ⁻⁶	0.005	2×10 ⁻⁶ (two per million)
1,2-Dichloroethane (1,2-DCA)	carcinogenicity (causes cancer)	0.0004	1×10 ⁻⁶	0.0005	1×10 ⁻⁶ (one per million)
1,1-Dichloroethylene (1,1-DCE)	hepatotoxicity (harms the liver)	0.01	NA	0.006	NA
1,2- Dichloroethylene, cis	nephrotoxicity (harms the kidney)	0.013	NA	0.006	NA
1,2- Dichloroethylene, trans	immunotoxicity (harms the immune system)	0.05	NA	0.01	NA
<u>Dichloromethane</u> (methylene chloride)	carcinogenicity (causes cancer)	0.004	1×10 ⁻⁶	0.005	1×10 ⁻⁶ (one per million)

Chemical Health Risk Categor	PHG R	Cancer Risk ³ at the PHG	Cancer Risk at the California MCL
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Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

2,4- Dichlorophenoxyacetic acid (2,4-D)	hepatotoxicity and nephrotoxicity (harms the liver and kidney)	0.02	NA	0.07	NA
1,2-Dichloropropane (propylene dichloride)	carcinogenicity (causes cancer)	0.0005	1×10 ⁻⁶	0.005	1×10 ⁻⁵ (one per hundred thousand)
1,3-Dichloropropene (Telone II□)	carcinogenicity (causes cancer)	0.0002	1×10 ⁻⁶	0.0005	2×10 ⁻⁶ (two per million)
<u>Dinoseb</u>	reproductive toxicity (harms the uterus and testis)	0.014	NA	0.007	NA
Diquat	ocular toxicity (harms the eye) developmental toxicity (causes malformation)	0.006	NA	0.02	NA
<u>Endothall</u>	digestive system toxicity (harms the stomach or intestine)	0.094	NA	0.1	NA
<u>Endrin</u>	neurotoxicity (causes convulsions) hepatotoxicity (harms the liver)	0.0003	NA	0.002	NA
Ethylbenzene (phenylethane)	hepatotoxicity (harms the liver)	0.3	NA	0.3	NA

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Ethylene dibromide	carcinogenicity	0.00001	1×10 ⁻⁶	0.00005	5×10 ⁻⁶
(1,2Dibromoethane)	(causes cancer)				(five per
					million)

Chemical	Health Risk Category ¹	California PHG (mg/L) ²	Cancer Risk³ at the PHG	California MCL ⁴ (mg/L)	Cancer Risk at the California MCL
<u>Fluoride</u>	musculoskeletal toxicity (causes tooth mottling)	1	NA	2	NA
<u>Glyphosate</u>	nephrotoxicity (harms the kidney)	0.9	NA	0.7	NA
<u>Heptachlor</u>	carcinogenicity (causes cancer)	0.00008 (8×10 ⁻⁶)	1×10 ⁻⁶	0.00001	1×10 ⁻⁶ (one per million)
Heptachlor epoxide	carcinogenicity (causes cancer)	0.00006 (6×10 ⁻⁶)	1×10 ⁻⁶	0.00001	2×10 ⁻⁶ (two per million)
Hexachlorobenzene	carcinogenicity (causes cancer)	0.00003	1×10 ⁻⁶	0.001	3×10 ⁻⁵ (three per hundred thousand)
Hexachlorocyclopentadiene (HCCPD)	digestive system toxicity (causes stomach lesions)	0.002	NA	0.05	NA

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Lead	developmental neurotoxicity (causes neurobehavioral effects in children) cardiovascular toxicity (causes high blood pressure) carcinogenicity (causes cancer)	0.0002	<1×10-6 (PHG is not based on this effect)	0.015 (AL ⁸)	2×10 ⁻⁶ (two per million)
<u>Lindane (γ-BHC)</u>	carcinogenicity (causes cancer)	0.000032	1×10 ⁻⁶	0.0002	6×10 ⁻⁶ (six per million)
Mercury (inorganic)	nephrotoxicity (harms the kidney)	0.0012	NA	0.002	NA
		California	Cancer	California	Cancer
Chemical	Health Risk Category ¹	PHG (mg/L) ²	Risk ³ at the PHG	MCL⁴ (mg/L)	Risk at the California MCL
Chemical Methoxychlor Methyl tertiarybutyl ether	endocrine toxicity (causes hormone effects) carcinogenicity (causes		the		California
Methoxychlor	endocrine toxicity (causes hormone effects)	(mg/L) ² 0.00009	the PHG NA	(mg/L) 0.03	California MCL NA

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Monochlorobenzene (chlorobenzene)	nephrotoxicity (harms the kidney)	0.07	NA	0.07	NA
<u>Nickel</u>	developmental toxicity (causes increased neonatal deaths)	0.012	NA	0.1	NA
<u>Nitrate</u>	hematotoxicity (causes methemoglobinemia)	45 as nitrate	NA	10 as nitrogen (=45 as nitrate)	NA
<u>Nitrite</u>	hematotoxicity (causes methemoglobinemia)	3 as nitrite	NA	1 as nitrogen (=3 as nitrite)	
Nitrate and Nitrite	hematotoxicity (causes methemoglobinemia)	10 as nitrogen ¹¹	NA	10 as nitrogen	NA

Chemical	Health Risk Category ¹	California PHG (mg/L) ²	Cancer Risk³ at the PHG	California MCL ⁴ (mg/L)	Cancer Risk at the California MCL
N-nitrosodimethyl- amine (NDMA)	carcinogenicity (causes cancer)	0.000003 (3×10 ⁻⁶)	1×10 ⁻⁶	none	NA

¹¹ The joint nitrate/nitrite PHG of 10 mg/L (10 ppm, expressed as nitrogen) does not replace the individual values, and the maximum contribution from nitrite should not exceed 1 mg/L nitrite-nitrogen.

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

<u>Oxamyl</u>	general toxicity	0.026	NA	0.05	NA
	(causes body weight effects)				
Pentachlorophenol (PCP)	carcinogenicity (causes cancer)	0.0003	1×10 ⁻⁶	0.001	3×10 ⁻⁶ (three per million)
<u>Perchlorate</u>	endocrine toxicity (affects the thyroid) developmental toxicity (causes neurodevelopmental deficits)	0.001	NA	0.006	NA
<u>Picloram</u>	hepatotoxicity (harms the liver)	0.166	NA	0.5	NA
Polychlorinated biphenyls (PCBs)	carcinogenicity (causes cancer)	0.00009	1×10 ⁻⁶	0.0005	6×10 ⁶ (six permillion)
Radium-226	carcinogenicity (causes cancer)	0.05 pCi/L	1×10 ⁻⁶	5 pCi/L (combined Ra226+228)	1×10 ⁻⁴ (one perten thousand)
Radium-228	carcinogenicity (causes cancer)	0.019 pCi/L	1×10 ⁻⁶	5 pCi/L (combined Ra226+228)	3×10 ⁻⁴ (three per ten thousand)
<u>Selenium</u>	integumentary toxicity (causes hair loss and nail damage)	0.03	NA	0.05	NA

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Chemical	Health Risk Category ¹	California PHG (mg/L) ²	Cancer Risk³ at the PHG	California MCL ⁴ (mg/L)	Cancer Risk at the California MCL
<u>Silvex (2,4,5-TP)</u>	hepatotoxicity (harms the liver)	0.003	NA	0.05	NA
<u>Simazine</u>	general toxicity (causes body weight effects)	0.004	NA	0.004	NA
Strontium-90	carcinogenicity (causes cancer)	0.35 pCi/L	1×10 ⁻⁶	8 pCi/L	2×10 ⁻⁵ (two perhundred thousand)
Styrene (vinylbenzene)	carcinogenicity (causes cancer)	0.0005	1×10 ⁻⁶	0.1	2×10 ⁻⁴ (two perten thousand)
1,1,2,2Tetrachloroethane	carcinogenicity (causes cancer)	0.0001	1×10 ⁻⁶	0.001	1×10 ⁻⁵ (one per hundred thousand)
2,3,7,8- Tetrachlorodibenzo- pdioxin (TCDD, or dioxin)	carcinogenicity (causes cancer)	5×10 ⁻¹¹	1×10 ⁻⁶	3×10 ⁻⁸	6×10 ⁻⁴ (six perten thousand)
Tetrachloroethylene (perchloroethylene, or PCE)	carcinogenicity (causes cancer)	0.00006	1×10 ⁻⁶	0.005	8×10 ⁻⁵ (eight per hundred thousand)

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<u>Thallium</u>	integumentary toxicity	0.0001	NA	0.002	NA
	(causes hair loss)				

		California	Cancer	California	Cancer
Chemical	Hoolth Biok Cotogony	PHG	Risk³ at the	MCL⁴	Risk at the
Chemicai	Health Risk Category ¹	(mg/L) ²	PHG	(mg/L)	California
					MCL
<u>Thiobencarb</u>	general toxicity	0.042	NA	0.07	NA
	(causes body weight effects)				
	hematotoxicity				
	(affects red blood cells)				
<u>Toluene</u>	hepatotoxicity	0.15	NA	0.15	NA
(methylbenzene)	(harms the liver) endocrine toxicity				
	(harms the thymus)				
<u>Toxaphene</u>	carcinogenicity (causes cancer)	0.00003	1×10 ⁻⁶	0.003	1×10 ⁻⁴ (one perten thousand)
1,2,4- Trichlorobenzene	endocrine toxicity (harms adrenal glands)	0.005	NA	0.005	NA
1,1,1- Trichloroethane	neurotoxicity (harms the nervous system), reproductive toxicity (causes fewer offspring) hepatotoxicity (harms the liver) hematotoxicity	1	NA	0.2	NA

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	(causes blood effects)				
1,1,2- Trichloroethane	carcinogenicity (causes cancer)	0.0003	1x10 ⁻⁶	0.005	2×10 ⁵ (two per hundred thousand)
Trichloroethylene (TCE)	carcinogenicity (causes cancer)	0.0017	1×10 ⁻⁶	0.005	3×10 ⁻⁶ (three per million)

Chemical	Health Risk Category¹	California PHG (mg/L) ²	Cancer Risk³ at the PHG	California MCL ⁴ (mg/L)	Cancer Risk at the California MCL
Trichlorofluoromethane (Freon 11)	accelerated mortality (increase in early death)	1.3	NA	0.15	NA
1,2,3-Trichloropropane (1,2,3-TCP)	carcinogenicity (causes cancer)	0.000007 (7×10 ⁻⁷)	1x10 ⁻⁶	0.000005 (5×10 ⁻⁶)	7×10 ⁻⁶ (seven per million)
1,1,2-Trichloro1,2,2- trifluoroethane (Freon 113)	hepatotoxicity (harms the liver)	4	NA	1.2	NA

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Tritium	carcinogenicity (causes cancer)	400 pCi/L	1x10 ⁻⁶	20,000 pCi/L	5x10 ⁻⁵ (five per hundred thousand)
<u>Uranium</u>	carcinogenicity (causes cancer)	0.43 pCi/L	1×10 ⁻⁶	20 pCi/L	5×10 ⁻⁵ (five per hundred thousand)
<u>Vinyl chloride</u>	carcinogenicity (causes cancer)	0.00005	1×10 ⁻⁶	0.0005	1×10 ⁻⁵ (one per hundred thousand)
<u>Xylene</u>	neurotoxicity (affects the senses, mood, and motor control)	1.8 (single isomer or sum of isomers)	NA	1.75 (single isomer or sum of isomers)	NA

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

		US EPA	Cancer	California	Cancer
Chemical	Health Risk Category ¹²	MCLG ¹³ (mg/L)	Risk ¹⁴ @ MCLG	MCL ¹⁵ (mg/L)	Risk @ California MCL
Disinfection byprod	lucts (DBPs)				
Chloramines	acute toxicity (causes irritation) digestive system toxicity (harms the stomach) hematotoxicity (causes anemia)	4 16,17	NA ¹⁸	none	NA
Chlorine	acute toxicity (causes irritation) digestive system toxicity (harms the stomach)	4 5,6	NA	none	NA
Chlorine dioxide	hematotoxicity (causes anemia) neurotoxicity (harms the nervous system)	0.85,6	NA	none	NA
Disinfection byprod	system) ucts: haloacetic acids (HAA	5)			

¹² Health risk category based on the US EPA MCLG document or California MCL document unless otherwise specified.

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¹³ MCLG = maximum contaminant level goal established by US EPA.

 $^{^{14}}$ Cancer Risk = Upper estimate of excess cancer risk from lifetime exposure. Actual cancer risk may be lower or zero. 1×10^{-6} means one excess cancer case per million people exposed.

¹⁵ California MCL = maximum contaminant level established by California.

¹⁶ Maximum Residual Disinfectant Level Goal, or MRDLG.

¹⁷ The federal Maximum Residual Disinfectant Level (MRDL), or highest level of disinfectant allowed in drinking water, is the same value for this chemical.

¹⁸ NA = not available.

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Monochloroacetic acid (MCA)	general toxicity (causes body and organ	0.07	NA	none	NA
	weight changes ¹⁹)				
Dichloroacetic acid (DCA)	carcinogenicity (causes cancer)	0	0	none	NA

Chemical	Health Risk Category ¹	US EPA MCLG ² (mg/L)	Cancer Risk³ @ MCLG	California MCL ⁴ (mg/L)	Cancer Risk @ California MCL	
Trichloroacetic acid (TCA)	hepatotoxicity (harms the liver)	0.02	NA	none	NA	
Monobromoacetic acid (MBA)	NA	none	NA	none	NA	
Dibromoacetic acid (DBA)	NA	none	NA	none	NA	
Total haloacetic acids (sum of MCA, DCA, TCA, MBA, and DBA)	general toxicity, hepatotoxicity and carcinogenicity (causes body and organ weight changes, harms the liver and causes cancer)	none	NA	0.06	NA	
Disinfection byproducts: trihalomethanes (THMs)						
Bromodichloromethane (BDCM)	carcinogenicity (causes cancer)	0	0	none	NA	

 $^{^{\}rm 19}$ Body weight effects are an indicator of general toxicity in animal studies.

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Bromoform	carcinogenicity (causes	0	0	none	NA
	cancer)				
Chloroform	hepatotoxicity and nephrotoxicity	0.07	NA	none	NA
	(harms the liver and kidney)				
Dibromochloromethane (DBCM)	hepatotoxicity, nephrotoxicity, and neurotoxicity	0.06	NA	none	NA
	(harms the liver, kidney, and nervous system)				
		US EPA	Cancer	California	Cancer
Chemical		MCLG ²		MCL⁴	Risk @
	Health Risk Category ¹	(mg/L)		(mg/L)	California MCL
Total trihalomethanes	carcinogenicity	none	NA	0.08	NA
/cum of DDCM					
(sum of BDCM, bromoform, chloroform and	(causes cancer), hepatotoxicity,				
bromoform, chloroform					

Table 1: Health Risk Categories and Cancer Risk Values for Chemicals with California Public Health Goals (PHGs)

Gross alpha particles ²⁰	carcinogenicity	0 (²²¹ Po	0	15 pCi/L ¹⁰	up to 1x10 ⁻³
	(causes cancer)	included)		(includes	(for 210 Po, the
				²²⁶ Ra but not	most potent
				radon and	alpha emitter
				uranium)	
Beta particles and	carcinogenicity	0 (²¹⁰ Pb	0	50 pCi/L	up to 2x10 ⁻³
photon emitters ⁹	(causes cancer)	included)		(judged	(for 210 Pb, the
				equiv. to 4	most
				mrem/yr)	notont
				, ,.,	potent
					betaemitter)

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²⁰ MCLs for gross alpha and beta particles are screening standards for a group of radionuclides. Corresponding PHGs were not developed for gross alpha and beta particles. See the OEHHA memoranda discussing the cancer risks at these MCLs at http://www.oehha.ca.gov/water/reports/grossab.html.

²¹ pCi/L = picocuries per liter of water.